

Acta Crystallographica Section E

Structure Reports

Online

ISSN 1600-5368

Bis[(1-methyl-1H-tetrazol-5-yl)sulfanyl]-methane

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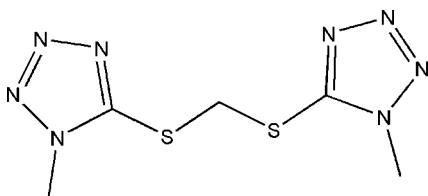
Received 23 March 2011; accepted 24 March 2011

Key indicators: single-crystal X-ray study; $T = 296$ K; mean $\sigma(\text{N}-\text{C}) = 0.007$ Å; R factor = 0.063; wR factor = 0.108; data-to-parameter ratio = 13.4.

The molecule of the title compound, $\text{C}_5\text{H}_8\text{N}_8\text{S}_2$, lies on a twofold rotation axis that relates on 1-methyltetrazolyl group to the other; the five-membered rings are twisted by 53.1 (1)°.

Related literature

For the synthesis and pharmacological activity of compounds containing tetrazole groups, see: Semenov (2002); Upadhayaya *et al.* (2004). For a related structure, see: Bronisz (2002).



Experimental

Crystal data

$\text{C}_5\text{H}_8\text{N}_8\text{S}_2$	$V = 1041.9$ (7) Å ³
$M_r = 244.31$	$Z = 4$
Orthorhombic, $Pbcn$	Mo $K\alpha$ radiation
$a = 6.415$ (3) Å	$\mu = 0.49$ mm ⁻¹
$b = 7.314$ (3) Å	$T = 296$ K
$c = 22.204$ (8) Å	$0.15 \times 0.12 \times 0.08$ mm

Data collection

CBruker SMART area-detector diffractometer	4692 measured reflections
Absorption correction: multi-scan (SADABS; Bruker, 2002)	936 independent reflections
$T_{\min} = 0.930$, $T_{\max} = 0.962$	482 reflections with $I > 2\sigma(I)$
	$R_{\text{int}} = 0.118$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.063$	70 parameters
$wR(F^2) = 0.108$	H-atom parameters constrained
$S = 1.21$	$\Delta\rho_{\text{max}} = 0.34$ e Å ⁻³
936 reflections	$\Delta\rho_{\text{min}} = -0.38$ e Å ⁻³

Data collection: SMART (Bruker, 2002); cell refinement: SAINT (Bruker, 2002); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 2008); program(s) used to refine structure: SHELXL97 (Sheldrick, 2008); molecular graphics: SHELXTL (Sheldrick, 2008); software used to prepare material for publication: SHELXL97.

We gratefully acknowledge the National Science Foundation of China (No. 20873100) and the Natural Science Foundation of Shaanxi Province (No. SJ08B09).

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: NG5140).

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supplementary materials

Acta Cryst. (2011). E67, o999 [doi:10.1107/S1600536811011007]

Bis[(1-methyl-1*H*-tetrazol-5-yl)sulfanyl]methane

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Experimental

Sodium hydroxide (1.7 g, 0.043 mol) was added to 5-mercapto-1-methyltetrazole (5 g, 0.043 mol) in dry dimethylsulfoxide (35 ml). The reaction mixture was stirred at 363 K for 1 h. Dichloromethane (3.1 ml, 0.0215 mol) was then added to the solution dropwise with the formation of a grey suspension. The suspension was stirred for 4 h, cooled to room temperature and filtered. The solvent was removed completely under reduced pressure. The residue was recrystallized from ethanol to give a white crystalline product (2.94 g; m.p. 353 - 354 K). Single crystals of the title compound suitable for X-ray diffraction analysis were isolated after a week from a solution in acetone.

Refinement

All H atoms were positioned geometrically (C—H = 0.96 Å for aromatic CH₃ and 0.97 Å for CH₂ groups, respectively) and constrained to ride on their parent atoms with $U_{\text{iso}}(\text{H})$ values set to be -1.5 of the carrier atom.

Figures

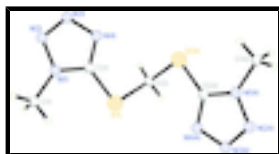


Fig. 1. A view of the molecular structure of title compound.

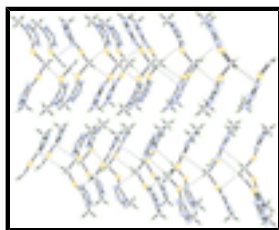


Fig. 2. The crystal packing of the title compound.

Bis[(1-methyl-1*H*-tetrazol-5-yl)sulfanyl]methane

Crystal data

C₅H₈N₈S₂

$M_r = 244.31$

Orthorhombic, *Pbcn*

Hall symbol: -P 2n 2ab

$a = 6.415(3)$ Å

$F(000) = 504$

$D_x = 1.558$ Mg m⁻³

$D_m = 1.558$ Mg m⁻³

D_m measured by not measured

Mo $K\alpha$ radiation, $\lambda = 0.71073$ Å

Cell parameters from 214 reflections

$\theta = 2.5$ – 18.9°

supplementary materials

$b = 7.314 (3) \text{ \AA}$	$\mu = 0.49 \text{ mm}^{-1}$
$c = 22.204 (8) \text{ \AA}$	$T = 296 \text{ K}$
$V = 1041.9 (7) \text{ \AA}^3$	Flake-like, colourless
$Z = 4$	$0.15 \times 0.12 \times 0.08 \text{ mm}$

Data collection

CBruker SMART area-detector diffractometer	936 independent reflections
Radiation source: fine-focus sealed tube graphite	482 reflections with $I > 2\sigma(I)$
φ and ω scans	$R_{\text{int}} = 0.118$
Absorption correction: multi-scan (SADABS; Bruker, 2002)	$\theta_{\text{max}} = 25.1^\circ$, $\theta_{\text{min}} = 1.8^\circ$
$T_{\text{min}} = 0.930$, $T_{\text{max}} = 0.962$	$h = -7 \rightarrow 7$
4692 measured reflections	$k = -8 \rightarrow 4$
	$l = -25 \rightarrow 26$

Refinement

Refinement on F^2	Primary atom site location: structure-invariant direct methods
Least-squares matrix: full	Secondary atom site location: difference Fourier map
$R[F^2 > 2\sigma(F^2)] = 0.063$	Hydrogen site location: inferred from neighbouring sites
$wR(F^2) = 0.108$	H-atom parameters constrained
$S = 1.21$	$w = 1/[\sigma^2(F_o^2) + (0.P)^2 + 0.7202P]$
936 reflections	where $P = (F_o^2 + 2F_c^2)/3$
70 parameters	$(\Delta/\sigma)_{\text{max}} < 0.001$
0 restraints	$\Delta\rho_{\text{max}} = 0.34 \text{ e \AA}^{-3}$
	$\Delta\rho_{\text{min}} = -0.38 \text{ e \AA}^{-3}$

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R -factors(gt) etc. and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	x	y	z	$U_{\text{iso}}^*/U_{\text{eq}}$	Occ. (<1)
S1	1.1330 (2)	1.03366 (18)	0.19301 (6)	0.0521 (4)	
N3	0.6134 (7)	0.9428 (6)	0.1121 (2)	0.0590 (13)	

C2	0.9188 (8)	0.9782 (6)	0.1491 (2)	0.0391 (13)	
N4	0.7207 (7)	1.0077 (6)	0.16107 (17)	0.0498 (12)	
N1	0.9358 (7)	0.8966 (6)	0.09509 (17)	0.0463 (11)	
C1	1.1179 (8)	0.8355 (7)	0.0621 (2)	0.0617 (16)	
H1A	1.1692	0.9336	0.0375	0.093*	
H1B	1.0808	0.7338	0.0369	0.093*	
H1C	1.2242	0.7985	0.0899	0.093*	
N2	0.7401 (8)	0.8743 (6)	0.07305 (18)	0.0561 (13)	
C3	1.0000	1.1654 (9)	0.2500	0.050 (2)	
H3A	1.1011	1.2439	0.2697	0.074*	0.50
H3B	0.8989	1.2439	0.2303	0.074*	0.50

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
S1	0.0481 (9)	0.0665 (10)	0.0418 (8)	0.0008 (8)	-0.0041 (7)	-0.0064 (7)
N3	0.049 (3)	0.061 (3)	0.068 (3)	-0.006 (3)	-0.010 (3)	0.001 (3)
C2	0.050 (4)	0.034 (3)	0.033 (3)	-0.004 (3)	-0.002 (2)	0.004 (2)
N4	0.041 (3)	0.061 (3)	0.047 (3)	0.002 (2)	0.004 (2)	0.004 (2)
N1	0.051 (3)	0.053 (3)	0.035 (2)	-0.004 (2)	-0.003 (2)	-0.003 (2)
C1	0.062 (4)	0.075 (4)	0.049 (3)	0.001 (3)	0.004 (3)	-0.012 (3)
N2	0.051 (3)	0.068 (3)	0.050 (3)	-0.004 (3)	-0.008 (3)	0.000 (2)
C3	0.061 (6)	0.054 (5)	0.034 (4)	0.000	-0.012 (4)	0.000

Geometric parameters (\AA , $^\circ$)

S1—C2	1.734 (5)	N1—C1	1.450 (6)
S1—C3	1.805 (4)	C1—H1A	0.9600
N3—N2	1.289 (5)	C1—H1B	0.9600
N3—N4	1.372 (5)	C1—H1C	0.9600
C2—N4	1.316 (6)	C3—S1 ⁱ	1.805 (4)
C2—N1	1.343 (5)	C3—H3A	0.9700
N1—N2	1.357 (5)	C3—H3B	0.9700
C2—S1—C3	98.31 (19)	H1A—C1—H1B	109.5
N2—N3—N4	110.6 (4)	N1—C1—H1C	109.5
N4—C2—N1	109.4 (4)	H1A—C1—H1C	109.5
N4—C2—S1	127.8 (4)	H1B—C1—H1C	109.5
N1—C2—S1	122.8 (4)	N3—N2—N1	107.1 (4)
C2—N4—N3	105.5 (4)	S1 ⁱ —C3—S1	115.5 (4)
C2—N1—N2	107.5 (4)	S1 ⁱ —C3—H3A	108.4
C2—N1—C1	130.8 (5)	S1—C3—H3A	108.4
N2—N1—C1	121.7 (4)	S1 ⁱ —C3—H3B	108.4
N1—C1—H1A	109.5	S1—C3—H3B	108.4
N1—C1—H1B	109.5	H3A—C3—H3B	107.5

Symmetry codes: (i) $-x+2, y, -z+1/2$.

Fig. 1

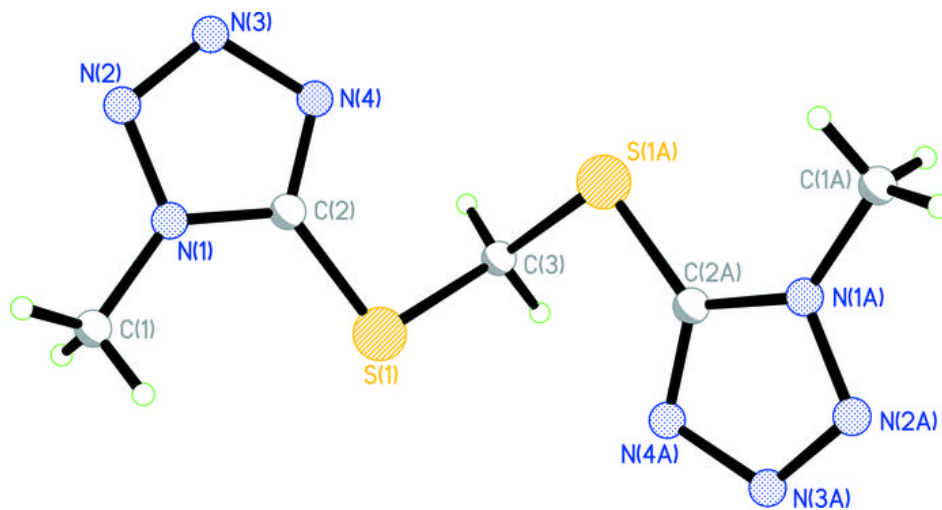


Fig. 2

